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Dissociation of Electron Transfer Dynamics and Product Characteristics of O₂ Reduction by FMO-containing Photosystem II Complexes. It was demonstrated earlier that the fraction of quenched chlorophyll triplet acceptor (~50%) in Photosystem II (PSII) is decreasing upon the action of an external charge transfer promoter, guanidine hydrochloride (GuHCl), due to a decrease in the electron transfer rate constant. Here it is shown that PSII complex dissociation leads to a more rapid electron transfer from excited PSII chlorophylls in the presence of GuHCl. Three different forms of PSII labeled with methylcarbonyl quencher of Chl fluorescence, 2,6-dichlorophenolindophenol (2,6-DCPIP), were used: (i) chlorophyll-deficient PSII (core), (ii) PSII in the presence of F(X) and F(A) (reconstituted) with fluorescent proteins separated by gel electrophoresis, and (iii) isolated PSII complex with the fluorescent photosynthetic reaction center (RC) (native).

The fluorescence of RC was excited at 550 nm in the presence of the probe 2,6-DCPIP and the sample was excited at 530 nm using low-intensity light. In the presence of GuHCl, an increased quantum yield and lifetime of the QA(-) charge-separated state along with the red shift of the redox charge-separated state of 2,6-DCPIP in PSII in the presence of GuHCl were demonstrated. The influence of the external charge transfer promoter on the energy transfer from the antennae of PSII to the reaction center was investigated using PSII isolated from a cyanobacterium *Synechococcus* PCC 7942. The photocurrent measurements showed that the presence of GuHCl decreased the electron transfer rate constant, probably due to the dissociation of the PSII complex. This new method was used for the study of the effect of the electron transfer promoter on

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